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Evaluation of a bacterial algal control agent in tank-based experiments

M. Schmack, J. Chambers, S. Dallas

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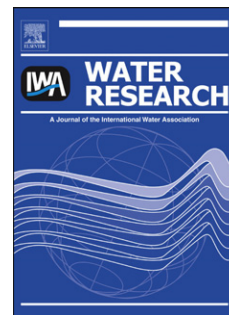
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1 Evaluation of a Bacterial Algal Control Agent in Tank-based Experiments

2

3 Schmack, M., Chambers, J. and Dallas, S., Murdoch University, Perth

4 South Street, Murdoch, Western Australia 6150

5 Phone Number: 61 (08) 9360 6399

6 E-mail: M.Schmack@murdoch.edu.au

7

8 **Abstract**

9 A bacterial-based bioremediation product, LakeRelief™ by Novozymes (Waterguru
10 LakeRelief 2011), was tested in a series of experiments between October 2008 and March
11 2009 to evaluate its suitability as a short-term intervention technique to reduce algal blooms
12 in the Swan-Canning River system. Results from fibreglass tank experiments (1100L)
13 suggested that the product did not actively attack and lyse algal cells. The product decreased
14 NH₄ and NO_x concentrations in treated tanks, both aerated and non-aerated. Product
15 application decreased PO₄ concentrations in non-aerated tanks but not in aerated tanks. The
16 product appeared to suppress algal growth in non-aerated tanks over short periods (several
17 days). Algal growth regularly diminished after product application but reappeared shortly
18 afterwards. Aeration had a negative effect on bacterial proliferation in the tanks, possibly
19 through alteration of environmental conditions (e.g. water mixing). As a consequence of the
20 environmental conditions in the tanks being counterproductive to the development of a
21 representative microbial composition, several aspects regarding the product's effectiveness
22 could not be assessed satisfactorily in the tank experiments. The importance of long-term
23 nutrient immobilisation into a well developed food web and the subsequent nutrient removal
24 through removal of the top order organisms is highlighted.

25

26 Keywords: Lake Relief, Bioremediation, Biological algal control, Algal lysing, Nutrient
27 competition, Bloom prevention

28

29 1. Introduction

30 Algal blooms are part of the normal seasonal development in aquatic ecosystems, (Sigeo
31 2005). In recent times, the use of fertilisers has dramatically accelerated eutrophication in
32 aquatic systems (Carpenter 2005). In the Swan-Canning River system, Western Australia,
33 long-term anthropogenic nutrient input has led to significant sediment enrichment,
34 particularly in phosphorus (Swan River Trust 2005) The phosphorus can then be released
35 from the river sediment under favourable P-release conditions (Xie, Xie, and Tang 2003) and
36 becomes rapidly available for algal growth. Consequently, harmful algae bloom (HAB)
37 outbreaks are symptomatic of the deteriorating health of the Swan-Canning River system.

38

39 A range of short-term intervention strategies have been trialled to reduce the frequency and
40 severity of algal blooms but with limited success. Past methods of algal control included the
41 use of algicides such as copper sulphate (Mason 1996). Several drawbacks from this method
42 were its non-selective nature, the resulting destruction of other freshwater biota, and its
43 prolonged presence in the water body. Other approaches that are more ecologically agreeable
44 are artificial oxidation (Kim 2006; Swan River Trust 2000) or application of clay-based
45 materials that bind phosphorus making it unavailable for algal growth (Pan et al. 2006; Swan
46 River Trust 2001). While each of these methods is valuable, a mechanism to rapidly remove
47 existing blooms is always attractive to managers dealing with immediate community concern.

48

49 An alternative approach to algal control is known as bioremediation or biological control.
50 Here, specially selected bacterial strains are introduced, they quickly populate the treated

51 water body and take up excess phosphate and nitrogen, thereby out-competing algal
52 populations (Kirchman 1994). Some bacteria in these products also excrete enzymes that
53 break down algae cell walls, a process that is known as algal lysing (Lee et al. 2000).
54 Bioremediation techniques are considered to have relatively low environmental impact and
55 are said to be free of unintended side effects such as fish kills or degradation of plant life
56 (Head 1998). Moreover, they attempt to stimulate or augment existing bacterial populations
57 in a water body but not to replace them. It is further claimed, but not scientifically verified,
58 that if these bio-control products are released into the environment, the introduced bacteria
59 will have no selective advantage over indigenous microbes, unless the targeted pollutants are
60 present (Waterguru Bertram Case Study 2011). However, attempts to clearly define the
61 efficacy of these products are made more difficult by the confidentiality of the active
62 constituents for various commercial-in-confidence reasons (Head 1998; Wang, Li, and Kang
63 2007).

64

65 Lake ReliefTM by Novozymes is a biological algae control product that has been used during
66 the last couple of years in field trials at Lake Bertram in Perth's southern suburbs and at the
67 Joondalup golf course, Perth, Western Australia (Waterguru Case Studies 2011). Anecdotal
68 findings from these trials suggest that the product has successfully reduced algal growth at
69 these locations. Moreover, it is claimed by the product distributor that treatment with the
70 product leads to a rapid disappearance of the surface scum in the treated water bodies, often
71 within a few days (Lee, personal communication 01/10/2008) However, no independent
72 scientific verification on the mechanism by which the product reduces algal growth has been
73 conducted.

74

75 Only limited information is publically available regarding the active ingredients in
76 LakeRelief. The product is described as a mixture of six different bacterial cultures with more
77 than 10^{12} microorganisms contained per 11b bag (Bjørn 2007). Due to these commercial-in-
78 confidence restrictions, the present study took an ecological approach to determine the
79 efficiency of LakeRelief. Owing to the potential ecological effects of product application into
80 open water bodies, it was decided to conduct the study in mesocosms. This led to the design
81 of several tank based experiments with the aim to verify if the product was able to reduce
82 algal blooms and by which mechanism - whether by lysing algal cells, a requirement essential
83 for the rapid disappearance of algal blooms within a few days as reported by the water
84 manager, or via the reduction of nutrient concentrations, thus out-competing algae for vital
85 nutrients. If LakeRelief was able to reduce or minimise the potential for algal blooms and
86 effectively contribute to algal bloom decline without any adverse ecological impact, the
87 product could play a significant role in improving the health of rivers and wetlands, in the
88 Swan-Canning River system and elsewhere.

89

90 2. Methods

91 2.1. Investigating LakeRelief's algal lysing properties

92 The rapid disappearance of surface algal conglomerations within a few days, as described by
93 the water manager, suggested that algal cells might be actively attacked by algicidal bacteria
94 strains. These bacteria are capable of lysing algal cells on contact (Mayali and Doucette
95 2002). In order to verify the observations reported from the field and to assess LakeRelief's
96 capability for algal lysing, sixteen fibreglass tanks (1000mm height; 1800mm internal
97 diameter) located at Murdoch University were selected for this experiment. The tanks had
98 been filled with rain water during winter and in spring a diverse algal community had

99 developed naturally with a species composition typical of wetlands and rivers in the area.
100 Advantage was taken of this well-developed algal bloom of the type the product was
101 designed to reduce. *Oocystis spp.* (green algae) was the dominant algal species existing in the
102 tanks. The tanks also contained *Planktolyngbya subtilis* (blue-green algae), *Scenedesmus spp.*
103 (green algae), *Synechocystis spp.* (blue green algae) and traces of numerous other algal
104 species. Strongly elevated pH of 10.1 and dissolved oxygen (DO) concentrations above 140%
105 saturation (11.5mg/L) in the tanks indicated substantial photosynthetic activity. At the start of
106 the experiment as signified by the input of LakeRelief, orthophosphate (PO_4^-) concentrations
107 in the tanks were below 35 $\mu\text{g/L}$ and ammonium (NH_4^-) concentrations were below 20 $\mu\text{g/L}$.
108 While both values were well within the range of healthy wetlands (ANZECC guidelines) and
109 given that this experiment was solely aimed at testing the immediate effect of product
110 bacteria on a well-established algal community, no additional nutrients were added to the
111 tanks. Please note: ANZECC water quality guidelines were used throughout this study to
112 reference conditions in the experimental tanks to those of natural systems in the region.

113

114 Due to the rapid growth of oxygen consuming bacteria and the potential ecological
115 consequences of deoxygenation, it was recommended to aerate treated water bodies when
116 using the product in the field (Lee, personal communication 01/10/2008). The impacts of the
117 product under aerated and non-aerated conditions were therefore investigated. The treatments
118 of 'product' and 'aeration' resulted in a factorial experiment with four groups each with four
119 repeats (1. product/aeration; 2. product/no aeration; 3. control/aeration; 4. control/no
120 aeration). The recommended product dosage for field application was 2ppm (Lee, personal
121 communication 01/10/2008). However, in order to observe the potential effect of a reduced
122 dosage, it was decided to begin the experiment with a product application rate of 1ppm.
123 Coinciding with the first product application, aeration was initiated in the corresponding

124 tanks (Precision SR 2500 aquarium air pumps). After three days and in the absence of an
125 immediate response, the corresponding tanks were re-dosed at the higher dose of 2ppm. Over
126 a period of one week, sets of water quality parameters (see Chapter 2.4) were analysed daily.

127

128 2.2. Investigating nutrient competition and algal growth suppression

129 by LakeRelief

130 In order to evaluate the product's ability to out-compete algal organisms for the
131 macronutrients nitrogen and phosphorus, a test was designed that simulated an unpolluted
132 water body with little microbiological activity that suddenly experienced high nutrient
133 pollution. The underlying theory was that product bacteria would rapidly multiply and
134 incorporate the macronutrients phosphorus and nitrogen into bacterial biomass and thus,
135 render them unavailable for algal growth (Doucette 1995; Doucette et al. 1998). Nutrient
136 input as a trigger for microbial growth and product addition occurred simultaneously, thus
137 allowing for an assessment of the product's ability to immediately out-compete and
138 consequently, suppress or reduce algal bloom development.

139

140 For this experiment, the tanks were drained, cleaned, and refilled with 1100 litres of
141 groundwater from the Murdoch veterinary farm at Murdoch University. The tanks were then
142 left to settle for one week. Based on visual observation, no significant algal growth occurred
143 during this period. In order to seed the tanks with river-specific algal cells, river sediment and
144 water were collected from the Canning River (Western Australia), approximately 150 metres
145 upstream from the Kent Street Weir. This location was chosen as it commonly has algal
146 blooms in summer and autumn (Swan River Trust 2005). Approximately one kilogram of
147 sediment and five litres of river water were then added to each tank.

148

149 After a one day settling period, all tanks were fertilized with 'Thrive soluble all purpose plant
150 food' by Yates to a calculated total phosphorus (TP) concentration of 2mg/L and a calculated
151 total nitrogen (TN) concentration of 10mg/L of tank water. These nutrient concentrations
152 represented highly eutrophic conditions that would result in immediate and rapid algal
153 growth. Aeration was initiated in the corresponding tanks and maintained over the duration of
154 the experiment. The tanks selected for 'product' treatment were then supplied with 4ppm
155 product. This higher dosage was chosen based on previous observations in the algal lysing
156 experiment, where routine nutrient monitoring had not demonstrated a significant effect on
157 the phosphorus and nitrogen levels in response to LakeRelief application. When the initial
158 input did not enable an effect to be definitely observed, incremental doses were used
159 throughout the experiment to attempt to produce a significant effect (as indicated by red
160 arrows on Figures 4 to 9). Physiochemical parameters and chlorophyll *a* were measured twice
161 weekly and nutrients were analysed once a week. On several occasions, samples for the
162 microscopic identification of algal species were taken from the tanks and analysed, so as to
163 monitor changes in algal composition over time. The experiment was conducted over a period
164 of seven weeks.

165

166 2.3. Confirmation of product viability

167 Prior to each application into the corresponding tanks, the product was dissolved in tap water.
168 For this purpose, the tap water had previously been aerated for twenty-four hours to facilitate
169 the removal of chlorine, which could otherwise have impacted negatively on bacterial health.
170 In order to verify the vitality of the product constituents, bacterial cultures were grown on
171 DifcoTM nutrient agar plates (Becton, Dickinson and Company). A small amount of the
172 product (approximately 20µg) was dissolved in 5ml sterile isotonic NaCl solution. This

173 emulsion was inoculated in concentrated form onto one plate whereas a second plate received
174 a 1:10 dilution.

175

176 2.4. Analysis

177 Nutrient analyses (TP, PO₄, TN, NH₄, NO_x) were carried out by the Marine and Freshwater
178 Research Laboratory (MAFRL; NATA accredited No. 10603) at Murdoch University.
179 Filterable reactive phosphorus was analysed by the ascorbic acid method (Johnson 1982);
180 nitrate plus nitrite by copper-cadmium reduction (Johnson 1983) and ammonium by the
181 alkaline phenate method (Switala 1993). Total nitrogen and total phosphorus were
182 determined from autoclave digests with potassium persulphate (Valderrama 1981). Sediment
183 samples were digested in concentrated sulphuric acid in the presence of a copper catalyst for
184 total Kjeldahl nitrogen (TKN) and total Kjeldahl phosphorus (TKP). All analyses were
185 carried out on a Lachat Quick-Chem 8000 Automated Flow Injection Analyser. The
186 remaining parameters were determined either on-site (turbidity, pH, DO, temperature,
187 conductivity), at the Loneragan laboratory facility at Murdoch University (COD), or at the
188 MAFRL laboratory facility at Murdoch University (chlorophyll *a* by fluorescence method).
189 The equipment used for these analyses is listed below (Table 1).

190

191 Table 1: Equipment list for physiochemical water analysis (used in all experiments in this
192 study)

193

194 Statistical analysis was carried out via SPSS 16 for Windows (Statistical Product Service
195 Solutions, Chicago, IL, USA, 2007). Because each microcosm was measured repeatedly, F-
196 values and significance levels in the following results section were obtained from a statistical
197 method known as ‘doubly multivariate repeated measures design’. Interaction terms were

198 initially included in the analysis but did not produce significant results and were consequently
199 removed from the model. Relationships between parameters were examined by computing
200 Spearman rank-order correlation, a non-parametric test for data that is not normally
201 distributed (Dytham 2005).

202

203 3. Results

204 3.1. Investigating the vitality of LakeRelief bacteria

205 After an incubation period of twenty four hours at 30⁰C, the cultivations prepared on Difco™
206 nutrient agar plates demonstrated strong bacterial growth. Both plates (concentrated and 1:10
207 dilution) were densely colonised, suggesting that a large number of vital bacteria were
208 contained in the product.

209

210 3.2. Investigating LakeRelief's algal lysing properties

211 3.2.1. pH

212 The naturally developed and well established algal blooms had created extremely alkaline pH
213 conditions in the tanks. Measurements revealed a pH of 10.1, well above the ANZECC
214 threshold of 8.0 that is indicative of the ecological health in local wetlands (ANZECC 2000).
215 Coincidentally, the random selection of tanks into the four test groups had led to a higher
216 algal incidence in the two product groups (both aerated and non-aerated). In response to these
217 unequal amounts of initial algal biomass, the product groups overall demonstrated higher pH
218 compared to control groups. A large amount of the variation between sampling dates resulted
219 from the prevailing climatic conditions (i.e. pH as a function of light intensity). Statistically,
220 neither product ($F_{1,12} = 3.621$, $P = 0.081$) nor aeration ($F_{1,12} = 0.000$, $P = 0.983$)
221 significantly affected pH. However, Figure 1 demonstrates that while pH in product/aeration

222 tanks behaved similarly to the pH in control groups, the pH in the product/no aeration group
223 was slightly reduced in response to product application. This suggested that algal metabolism
224 was somewhat diminished in this group.

225

226 Figure 1: Testing algal lysing properties - pH value of the water column in product and
227 control tanks (P/A = microcosms with product and aeration, P/na = microcosms with product
228 and no aeration, c/A = microcosms with no product (control) and aeration, c/na = microcosms
229 with no product (control) and no aeration). Red arrows indicate product application.

230

231 3.2.2. Dissolved Oxygen (DO)

232 DO was supersaturated in all tanks and exceeded the ANZECC threshold of 120% saturation
233 (10mg/L), reflecting strong algal photosynthetic activity (Figure 2). The different DO
234 concentrations between the groups signify the initial unequal presence of algal biomass and
235 the variation between the sampling dates represents climatic influences. The induction of
236 aeration immediately reduced DO in aerated tanks compared to non-aerated tanks. As DO
237 concentrations in aerated tanks remained lower throughout the experiment, this was
238 significant ($F_{1,12} = 8.369$, $P < 0.05$). In contrast, the second factor product did not produce a
239 significant effect on DO concentrations ($F_{1,12} = 2.765$, $P = 0.122$).

240

241 Figure 2: Testing algal lysing properties - DO saturation of the water column in product and
242 control tanks

243

244 3.2.3. Chlorophyll *a*

245 The unequal biomass distribution between the four test groups was once again demonstrated
246 in the chlorophyll *a* concentrations at the start of the experiment (Figure 3). Regardless,

247 chlorophyll *a* concentrations in all groups exceeded trigger values of 3-5µg/L (ANZECC
248 2000). Due to large standard deviations, both factors product ($F_{1,12} = 3.252$, $P = 0.096$) and
249 aeration did not significantly affect chlorophyll *a* concentrations ($F_{1,12} = 1.175$, $P = 0.300$).
250 However, while chlorophyll *a* concentration increased in the product/aeration tanks at a rate
251 comparable to control tanks, it reduced over time in the product/no aeration group, suggesting
252 reduced algal metabolism.

253

254 Figure 3: Testing algal lysing properties - Chlorophyll *a* concentration of the water column in
255 product and control tanks

256

257 3.3. Investigating nutrient competition by LakeRelief

258 3.3.1. Orthophosphate (PO_4)

259 PO_4 concentrations in the tanks were initially between 30 and 50 µg/L. Following nutrient
260 input, they strongly increased over the next four days in product/aeration and in both control
261 groups but remained much lower in product/no aeration (Figure 4). Nevertheless, the
262 ANZECC threshold of 40µg/L was dramatically exceeded in all groups. Although the effect
263 in the product/no aeration group was strongly masked by the dissimilar reaction in
264 product/aeration tanks, the statistical analysis of product versus control groups produced an
265 almost significant result. The measurable effect of factor product was just outside the 5%
266 significance level ($F_{1,12} = 4.252$, $P = 0.062$). Based on the large variation within groups,
267 aeration was not significant ($F_{1,12} = 1.601$, $P = 0.230$).

268

269 Figure 4: Testing nutrient competition - Orthophosphate concentration of the water column in
270 product and control tanks.

271

272 3.3.2. Ammonium (NH₄)

273 While less than 35 µg/L NH₄ was initially found in the water column (twenty-four hours
274 after massive nitrogen input as NH₄), all groups strongly exceeded the threshold value of
275 80µg/L (ANZECC 2000) shortly afterwards (Figure 5). The subsequent development of NH₄
276 concentrations was significantly determined by the factor product. In product tanks, NH₄
277 concentrations decreased faster (after Jan 26th) compared to control tanks ($F_{1,12} = 6.672$, $P <$
278 0.05). Deviating from the observations for orthophosphate, this response was demonstrated in
279 both product groups. After Feb 10th, NH₄ was almost completely absent from all product
280 tanks whereas around 200µg/L NH₄ were still available in control tanks. Aeration had no
281 significant effect on NH₄ concentrations ($F_{1,12} = 0.106$, $P = 0.750$).

282

283 Figure 5: Testing nutrient competition - Ammonium concentration of the water column in
284 product and control tanks.

285

286 3.3.3. Nitrite and Nitrate (NO_x)

287 The almost complete absence of NO_x (below 10µg/L) from all tanks twenty-four hours after
288 fertiliser input was followed by a dramatic increase over the next four days (Figure 6). After
289 NO_x had reached extreme concentrations (800% above the ANZECC threshold of 150µg/L),
290 their subsequent reduction was significantly accelerated by the product ($F_{1,12} = 10.177$, $P <$
291 0.01). On Feb 3rd, NO_x concentrations in product tanks were below 100µg/L, whereas they
292 were approximately five times higher in control tanks. After Feb 10th, NO_x was completely
293 absent from the product tanks, compared to around 200µg/L in control tanks. The factor
294 aeration was not significant ($F_{1,12} = 2.029$, $P = 0.180$).

295

296 Figure 6: Testing nutrient competition - Nitrite and Nitrate concentration of the water column
297 in product and control tanks

298

299 3.4. Investigating LakeRelief's ability to prevent, reduce or eliminate
300 algal blooms

301 3.4.1. pH

302 The immediate and extreme pH rise in response to algal seeding and nutrient addition reflects
303 the dramatic increase in algal metabolism in all groups (Figure 7). Within four days, algal
304 photosynthesis had resulted in extremely alkaline pH, dramatically exceeding ANZECC
305 guidelines of 6.5-8.0 (ANZECC 2000). A slightly faster and initially higher pH rise occurred
306 in non-aerated tanks. Furthermore, in aerated tanks pH fluctuation was less pronounced
307 throughout the experiment, somewhat moderated by aeration. Consequently, aeration had a
308 significant influence on pH ($F_{1,12} = 11.858$, $P < 0.01$). The pH in product/no aeration tanks
309 did often reduce when product bacteria were added (particularly after the second, fourth and
310 fifth dosing events), suggesting reduced algal photosynthesis. The same response could not
311 be observed for the product/aeration group, where the pH often increased after product
312 application (second, third and fourth dosing). The opposing reaction to product in this group
313 lead to a large variation within the two product groups and consequently, resulted in a non-
314 significant result for factor product overall ($F_{1,12} = 3.292$, $P = 0.095$).

315

316 Figure 7: Testing algal growth suppression - pH value of the water column in product and
317 control tanks

318

319 3.4.2. Dissolved Oxygen (DO)

320 Similar to pH, the DO increase in all groups immediately after nutrient input corresponded to
321 the dramatic increase in algal biomass and the intensified production of oxygen via
322 photosynthesis in all groups (Figure 8). However, this rise was significantly stronger in non-
323 aerated groups ($F_{1,12} = 12.752$, $P < 0.01$), where DO rapidly exceeded the threshold value of
324 120% saturation (10mg/L; ANZECC 2000). Throughout the experiment, aerated tanks
325 maintained 'healthier' DO concentrations than non-aerated tanks. In aerated tanks, DO never
326 fell below 75% saturation (6.5mg/L). Equally, aeration prevented extreme DO-
327 supersaturation. Whereas DO reached up to 200% saturation (>15mg/L) in non-aerated tanks,
328 DO never rose above 160% saturation (12mg/L) in aerated tanks. Thus, aeration was
329 effectively 'buffering' DO concentrations. Similar to previous findings for pH, DO
330 concentrations in product/no aeration tanks appeared to reduce in correspondence to the input
331 of product (second, fourth and fifth dosing). As this response to product application was not
332 evident in the product/aeration group, the discrepancy within the two product groups led to a
333 non-significant result overall ($F_{1,12} = 1.697$, $P = 0.217$).

334

335 Figure 8: Testing algal growth suppression – DO saturation of the water column in product
336 and control tanks

337

338 3.4.3. Chlorophyll *a*

339 Shortly after nutrient input, chlorophyll *a* exceeded ANZECC guidelines (3–5µg/L) in all
340 tanks (Figure 9). The subsequent increase of chlorophyll *a* reflects the build up of algal
341 biomass. Chlorophyll *a* markedly dropped in both control groups around Feb 3rd. The
342 product/no aeration curve confirmed the previously observed reduction in algal growth in
343 response to the second, fourth and fifth product dosing events. As this response was again not
344 demonstrated in the product/aeration group, the large variation between the product groups

345 resulted in overall insignificance for the factor product ($F_{1,12} = 1.996$, $P = 0.183$). Similarly,
346 aeration did not have a significant influence on chlorophyll *a* concentrations ($F_{1,12} = 0.951$, P
347 $= 0.349$).

348

349 Figure 9: Testing algal growth suppression – Chlorophyll *a* concentration of the water
350 column in product and control tanks

351

352 4. Discussion

353 4.1. Investigating LakeRelief's algal lysing properties

354 The value of algicidal bacteria as a factor in the control of algal blooms and in limiting algal
355 biomass has been highlighted by a number of studies (Lovejoy, Bowman, and Hallegraeff
356 1998; Mitsutani and Yamasaki 2001; Wu, Xi, and Zhao 2002). Studies in eutrophic
357 stormwater detention ponds suggest that algicidal bacteria are a more effective factor in
358 bloom termination than nutrient depletion (Bunker 2004; Lewitus et al. 2004; Lewitus et al.
359 2003). In regards to the mode of operation of the tested product, previous observations from
360 field trials described the rapid disappearance of algal blooms (Waterguru Case Studies 2011).
361 These anecdotal findings suggest that the product might contain algicidal bacterial strain(s),
362 capable of breaking up algal conglomerations by actively lysing algal cells.

363

364 Algicidal activity is operationally defined as the induction of algal cell lysis within one to two
365 days following bacterial introduction into algal cultures (Mayali and Doucette 2002). If
366 LakeRelief was capable of lysing algal cells in the present experiment, this would have
367 quickly led to dramatic changes in the tanks, signified by a strong and immediate reduction of
368 the key indicators for algal photosynthetic activity, i.e. pH, DO and chlorophyll *a*. However,

369 such changes were not observed in the tanks (Figures 1, 2 and 3). This does not definitively
370 preclude the presence of algal lysing strains in LakeRelief, it is possible that conditions in the
371 tanks (high pH and DO) may have suppressed these strains.

372

373 4.2. Investigating nutrient competition by LakeRelief

374 A study on bacterial nutrient uptake in lake and sea water indicated that the amount of
375 inorganic phosphate taken up by bacteria can exceed 50% of the entire phosphate pool
376 (Kirchman 1994). This suggests that bacteria are strong competitors for phosphate under
377 favourable circumstances. The situation is more complex for nitrogen, where bacteria are
378 traditionally considered superior in competition for organic forms of nitrogen, while algae
379 appear to be superior competitors for inorganic nitrogen supplied as NH_4 (Løvvdal et al. 2008).
380 Consequently, nutrient competition between the bacteria and algae varies depending upon
381 which form of nitrogen, either organic or inorganic, is more readily available.

382

383 In the present experiment there was principal support for the product's ability to out-compete
384 algae for nitrogen and phosphorus:

- 385 1. the product accelerated NH_4 and NO_x reduction in all treated tanks, both aerated and
386 non-aerated; and
- 387 2. product application resulted in PO_4 reduction in product/no aeration tanks only.
388 Aeration had a negative effect on bacterial proliferation, possibly through alteration of
389 environmental conditions in the tanks.

390 However, visual observations and experimental data analysis suggested that nutrient cycling
391 processes occurred at an accelerated rate in the confined tanks. As a result, the nutrient
392 removal rate achieved by product bacteria could not be accurately quantified in this
393 experiment.

394

395 The accelerated reduction of NH_4 and NO_x concentrations that was observed in both product
396 treatments and the corresponding chlorophyll *a* measurements as an indicator for algal
397 metabolism strongly suggested that product bacteria were responsible for the nitrogen uptake
398 (Figures 5 and 6). Based on the corresponding reduction of NH_4 and NO_x after Jan 26th, the
399 implication was that the product may have supplemented both nitrifying and denitrifying
400 bacterial strains in the tanks (Blackburn et al. 1994).

401

402 Regarding phosphorus, the findings were more difficult to interpret. In product/no aeration
403 tanks, algal biomass measured as chlorophyll *a* closely responded to the application of the
404 product, diminishing strongly particularly after the high dosing rates on day 19 (Feb 3rd) and
405 day 28 (Feb 12th) (Figure 4). Reduced algal activity in this group was demonstrated by
406 several more parameters (turbidity, pH, DO, COD). Coinciding with these observations, PO_4
407 concentrations in the water column were markedly lower in the product/no aeration group
408 compared to all other groups. The combination of standing algal biomass reduction and a
409 reduction of PO_4 suggested that PO_4 had been taken up by product bacteria in this treatment
410 type. The situation was different for the product/aeration group. Here, PO_4 concentrations
411 were similar to those in control groups. Furthermore, the fluctuation in algal activity appeared
412 to be unrelated to product application. This suggested that the effect of aeration had interfered
413 and reduced bacterial ability to take up PO_4 and out-compete algae in this group.

414

415 4.3. Investigating LakeRelief's ability to prevent, reduce or eliminate 416 algal blooms

417 Visual observations and analytical results (turbidity, chlorophyll *a*) provided good indication
418 that algal growth was inhibited in product/no aeration tanks in response to LakeRelief input

419 (Figure 9). In one case, a DO reduction to 23% saturation (2mg/L) shortly after product
420 application reflected the intensive bacterial metabolism that occurred in the tank. However,
421 the algal population appeared to quickly recover within a few days. Once again, the
422 product/aeration group did not demonstrate this short-term algal suppression in response to
423 product application. This suggested that artificial aeration hindered the development of
424 product bacteria in the confined mesocosms, possible related to the strong vertical mixing
425 effect. While this was an unfortunate consequence of the experimental design, it must be
426 stressed that due to the potential of aerobic bacteria to quickly metabolise the available
427 oxygen, aeration is a specified requirement in bioremediation trials in natural water bodies.

428

429 An additional aspect that may have over time strongly influenced the product's performance
430 in the aerated tanks was the algal species composition that evolved in response to continuous
431 aeration. While two species of green algae *Scenedesmus* spp. and *Selenastrum* sp. 001
432 appeared to be favoured in non-aerated (= stagnant) tanks where they were detected in large
433 numbers, cyanobacteria *Synechocystis* spp. was represented stronger in aerated tanks
434 throughout the experiment. This might be related to the ability of bacteria to exhibit
435 buoyancy migration (Sigeo 2005), which would promote them over green algae as
436 competitors for the available nutrients in the stagnant (= stratified) tanks.

437

438 4.4. Limitations in tank based experiments

439 Findings in the present study were in line with Duvall and Anderson (2001) who found that
440 the conditions for observing the competition between algae and bacteria in the confined space
441 of small scale experiments might not be optimal. Duvall, Anderson and Goldman (2001)
442 reported little difference amongst microbial products and non-treated controls even in
443 reactors of 61,000L capacity. This was despite the fact that they found a significant increase

444 of bacterial populations at least in the short term. The apparent interference of aeration on the
445 ability of bacteria to take up nutrients in the present study led to a more general consideration
446 of limitations in tank based experiments (see also Table 2).

447

448 An initial limitation to bacterial growth may have occurred in the nutrient competition
449 experiment, when algal seed material, nutrients and product bacteria were simultaneously
450 added to the tanks at the first product dosing event. Here bacterial growth was prevented due
451 to the absence of organic carbon as substrate (Bratbak 1987; Corina, Bussaard, and Riegman
452 1998). Thus, algal cells took up the available nutrients and rapidly increased their numbers to
453 the point where further population size increase was limited by other factors (e.g. reduced
454 light intensity, expressed as turbidity). Over time and in response to the build-up of algal
455 biomass, algal photosynthesis and algal excretion processes provided the organic substrate
456 necessary for bacterial growth (Brown and Zeiler 1993). This limitation would not have
457 occurred in a natural system where organic carbon is usually available as detritus etc.

458

459 Regarding physiochemical parameters, pH and DO strongly influenced the outcome of the
460 experiments. Recommendations for the optimum conditions that ensure good biological
461 activity of product bacteria are a pH within the range of 5.5-8.5 and DO concentrations of 6-
462 9mg/l (Lee, personal communication 01/02/2009). Unfortunately, the limited size of the tanks
463 promoted extreme pH and DO concentrations (Figures 1, 2, 7 and 8) and the recommended
464 thresholds were often dramatically exceeded in the tanks (pH up to 11; DO up to 200%
465 saturation or 15mg/L). It is likely that the high pH in the tanks led to a competitive advantage
466 of algae over bacteria and therefore restricted bacterial proliferation. Resulting from the finite
467 space of the tanks, several more factors likely affected the outcome of the tank experiments.
468 These factors are briefly summarised in Table 2.

469

470 Table 2: Limitations in tank experiments and their consequences

471

472 **4.5. Long-term nutrient immobilisation**

473 As previously discussed, visual observations and experimental data analysis led to the
474 conclusion that nutrient cycling processes occurred at an accelerated rate in the confined
475 space of the experiment tanks. A key requirement for long-term immobilisation of
476 macronutrients nitrogen and phosphorus and thus, suppression of algal growth over any
477 substantial time, is the establishment of a well developed trophic cascade. In order to be
478 successfully employed as a long-term algal bloom management tool, optimum growth
479 conditions for bacterial metabolism, a well developed trophic cascade and regular harvesting
480 of the accumulated nutrients through removal of highest level organisms would be essential.
481 This suggests that there may be an opportunity for biological control agents such as
482 LakeRelief as a long-term algal bloom management tool in the aqua-cultural industry were
483 nutrients are passed through a food web and are ultimately removed through harvesting of the
484 top order organisms in the system e.g. prawns or fish.

485

486 In contrast to this, in natural wetland management efficient nutrient removal through
487 harvesting of the top order organisms does not take place. As a result of the continuous
488 nutrient recycling in such a system, bioremediation programs can only suppress algal bloom
489 development by constantly dosing and maintaining unnaturally high levels of bio-control
490 bacteria. This brings up another aspect regarding the impact of bio-control bacteria on food
491 web structure. In a simplified model it could be hypothesised that when bacterial numbers are
492 continuously kept at higher than natural levels due to constant external input, their predators
493 (heterotrophic nanoflagellates and microzooplankton) would respond to the increased food

494 availability by increasing their own numbers over time. This would eventually lead to larger
495 numbers of organisms in the next higher trophic level (zooplankton grazers) and so forth.
496 While the build-up of particular food web species to unnaturally large numbers may occur in
497 natural systems in response to the temporary release of bottom-up or top-down pressure, both
498 pressures quickly increase in response to this build up. Thus, population density of the
499 previously uncontrolled species will decrease in response to the strengthened control
500 mechanisms. Therefore, it could be argued that the constant addition of bacteria as a food
501 source would lessen the regulatory strength of bottom-up pressure and ultimately lead to
502 ecosystem destabilisation and a much more rapid turnover of nutrients.

503

504 5. Conclusion

505 In this ecological study, a series of tank based experiments was conducted to investigate the
506 efficiency and the modus operandi of a bacterial based bioremediation product, LakeRelief™
507 by Novozymes. The ability to actively attack and lyse algal cells - essential for the rapid
508 disappearance of algal blooms within 1-2 days - was not observed. LakeRelief's capacity to
509 compete with algae for vital macronutrients was indicated. Under suitable conditions, the
510 product bacteria grew and incorporated available nutrients nitrogen and phosphorus into
511 bacterial biomass. However, due to the accelerated rate of nutrient recycling in the tanks,
512 resulting from the absence of a significant trophic cascade and the restricted space of the
513 mesocosms, it could not be statistically confirmed that the bacteria were successfully out-
514 competing algal organisms for vital macronutrients over any significant time period.
515 Essential conditions for bacterial growth were not satisfied in the confined space of the tanks
516 and thus, frustrated the process of algal growth suppression. Amongst these confining factors
517 were carbon limitation, extreme physiochemical conditions, water motion from aeration and
518 unfavourable light conditions in the tanks. While application of bio-control agents may

519 provide very temporary reduction in algal blooms in natural wetlands, the key requirements -
 520 long-term nutrient immobilisation into an artificially amplified food web and the subsequent
 521 removal of nutrients through removal of the top order organisms - are not satisfied. The
 522 requirement to continually add product to retain an algal suppression effect could result in
 523 rapid turnover of nutrients and ecosystem destabilisation.

524

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529

530 7. References

- 531 ANZECC. 2000. Australian and New Zealand Guidelines for Fresh and Marine Water
 532 Quality (Chapters 1-7). www.deh.gov.au/water/quality/nwqms.
 533 Blackburn, T. H., N. D. Blackburn, K. Jensen, and N. Risgaard-Petersen. 1994. Simulation
 534 Model of the Coupling between Nitrification and Denitrification in a Freshwater
 535 Sediment. *Appl. Environ. Microbiol.* 60 (9):3089-3095.
 536 Bratbak, G. 1987. Carbon flow in an experimental microbial ecosystem. *Marine ecology.*
 537 *Progress series (Halstenbek)* 36:267.
 538 Brown, L. M. and Zeiler, K. G. 1993. Aquatic biomass and carbon dioxide trapping.
 539 *Energy Conversion and Management.* 34 (9-11):1005-1013.
 540 Bunker, K.N. 2004. A hydrologic assessment of two detention pond watersheds in an urban
 541 coastal landscape, The Graduate School, College of Charleston.
 542 Carpenter, S.R. 2005. Eutrophication of aquatic ecosystems: Bistability and soil phosphorus.
 543 *Proceedings of the National Academy of Sciences of the United States of America.*
 544 102 (29):10002-10005.
 545 Corina, P. D., P. Bussaard, and R. Riegman. 1998. Influence of bacteria on phytoplankton
 546 cell mortality with phosphorus or nitrogen as the algal-growth-limiting nutrient.
 547 *Aquatic Microbial Ecology* 14:271-280.
 548 Doucette, G.J., M. Kodama, S. Franca, and S. Gallacher. 1998. Bacterial interactions with
 549 harmful algal bloom species: bloom ecology, toxigenesis, and cytology. In
 550 *Physiological Ecology of Harmful Algal Blooms*, edited by D. M. Anderson, A. D.
 551 Cembella and G. M. Hallegraeff. Berlin: Springer-Verlag.
 552 Doucette, Gregory J. 1995. Interactions between bacteria and harmful algae: A review.
 553 *Natural Toxins* 3 (2):65-74.

- 554 Duvall, R.J., and L.W. Anderson. 2001. Laboratory and Greenhouse Studies of Microbial
 555 Products Used to Biologically Control Algae. *Journal of Aquatic Plant Management*
 556 39:95-98.
- 557 Duvall, R.J., L.W. Anderson, and C.R. Goldman. 2001. Pond Enclosure Evaluations of
 558 Microbial Products and Chemical Algicides Used in Lake Management. *Journal of*
 559 *Aquatic Plant Management* 39:99-106.
- 560 Dytham, C. 2005. *Choosing and Using Statistics: A Biologist's Guide*. Oxford: Blackwell
 561 Publishing.
- 562 Head, I. 1998. Bioremediation: towards a credible technology. *Microbiology* 144:599-608.
- 563 Johnson, K.S. 1982. Determination of phosphate in seawater by flow injection analysis with
 564 injection of reagent. *Anal Chem.* 54:1185-1187.
- 565 Johnson, K.S. 1983. Determination of nitrate and nitrite in seawater by flow injection
 566 analysis with injection of reagent. *Limnology and Oceanography.* 28 (6):1260-1266.
- 567 Kim, H.G. 2006. Mitigation and Controls of HABs. In *Ecological Studies Volume 189*
 568 *Ecology of Harmful Algae*, edited by E. Granéli and J. T. Turner. Berlin: Springer-
 569 Verlag.
- 570 Kirchman, D.L. 1994. The uptake of inorganic nutrients by heterotrophic bacteria. *Microbial*
 571 *Ecology* 28:255-271.
- 572 Lee, S.J., Y. Kim, H.G. Kim, G.M. Seo, J.H. Jeong, and Y.K. Hong. 2000. Algalytic activity
 573 of α -mannosidase on harmful marine microalgae. *Journal of Applied Phycology* 12
 574 (2):191-193.
- 575 Lewitus, A.J., K.C. Hayes, J.W. Kempton, L.J. Mason, S.B. Wilde, B.J. Williams, and J.L.
 576 Williams. 2004. Prevalence of raphidophyte blooms in South Carolina brackish ponds
 577 associated with housing and golf courses. In *Harmful Algae 2002. Florida Fish and*
 578 *Wildlife Conservation Commission and Intergovernmental Oceanographic*
 579 *Commission of UNESCO*, edited by K. A. Steidinger, J. H. Landsberg, C. R. Tomas
 580 and G. A. Vargo.
- 581 Lewitus, A.J., L.B. Schmidt, L.J. Mason, J.W. Kempton, S.B. Wilde, J.L. Wolny, B.J.
 582 Williams, K.C. Hayes, S.N. Hymel, C.J. Keppler, and A.H. Ringwood. 2003. Harmful
 583 algal blooms in South Carolina residential and golf course ponds. *Population &*
 584 *Environment* 24:387-413.
- 585 Løvdal, T., C. Eichner, H.P. Grossart, V. Carbonnel, L. Chou, and T.F. Thingstad. 2008.
 586 Competition for inorganic and organic forms of nitrogen and phosphorous between
 587 phytoplankton and bacteria during an *Emiliania huxleyi* spring bloom. *Biogeosciences*
 588 5 (2):371.
- 589 Lovejoy, C., J. P. Bowman, and G. M. Hallegraeff. 1998. Algicidal effects of a novel marine
 590 *Pseudoalteromonas* isolate (Class Proteobacteria, Gamma Subdivision) on harmful
 591 algal bloom species of the genera *Chattonella*, *Gymnodinium*, and *Heterosigma*.
 592 *Applied and Environmental Microbiology* 64 (8):2806-2813.
- 593 Mason, C. F. 1996. *Biology of Freshwater Pollution*. 3 ed. Essex: Longman.
- 594 Mayali, X., and G.J. Doucette. 2002. Microbial community interactions and population
 595 dynamics of an algicidal bacterium active against *Karenia brevis* (Dinophyceae).
 596 *Harmful Algae* 1 (3):277-293.
- 597 Mitsutani, A., and I. Yamasaki. 2001. Analysis of algicidal proteins of a diatom-lytic marine
 598 bacterium *Pseudoalteromonas* sp. strain 25 by two-dimensional electrophoresis.
 599 *Phycologia* 40 (3):286-291.
- 600 Pan, G., H. Zou, H. Chen, and X. Yuan. 2006. Removal of harmful cyanobacterial blooms in
 601 Taihu Lake using local soils. III. Factors affecting the removal efficiency and an in
 602 situ field experiment using chitosan-modified local soils. *Environmental Pollution*
 603 141:206-212.

- 604 Sigeo, D.C. 2005. *Freshwater Microbiology: biodiversity and dynamic interactions of micro-*
605 *organisms in the freshwater environment*: John Wiley & Sons Ltd: UK.
- 606 Swan River Trust. 2000. Oxygenating the Swan and Canning rivers. In *River Science*. Perth:
607 The Government of Western Australia.
- 608 ———. 2001. Phosphorus in the Canning – 1999-2000 Phoslock™ trials. In *River Science*.
609 Perth: The Government of Western Australia.
- 610 ———. 2005. Algal blooms in the Swan-Canning estuary: Patterns, causes and history. In
611 *River Science*. Perth: The Government of Western Australia.
- 612 Switala, K. 1993. *Determination of ammonia by flow injection analysis colorimetry*
613 *(dialysis)*. Latchet Instruments, Milwaukee, USA.
- 614 Valderrama, J. 1981. The simultaneous analysis of total nitrogen and total phosphorus in
615 natural waters. *Marine Chemistry*. 10:109-122.
- 616 Wang, L., J. Li, and W.L. Kang. 2007. Bioremediation of Eutrophicated Water by
617 *Acinetobacter calcoaceticus*. *Bulletin of Environmental Contamination and*
618 *Toxicology* 78:537-530.
- 619 Wu, G., Y. Xi, and Y.J. Zhao. 2002. The latest development of research on algae-lysing
620 bacteria [J]. *Research of Environmental Sciences* 15 (5):43-46.
- 621 Xie, L.Q., P. Xie, and H.J. Tang. 2003. Enhancement of dissolved phosphorus release from
622 sediment to lake water by *Microcystis* blooms—an enclosure experiment in a hyper-
623 eutrophic, subtropical Chinese lake. *Environmental Pollution* 122:391-399.
- 624
- 625 Online Resources:
- 626
- 627 Bjørn, C. 2007. Lake Relief™— effect on filamentous algae, nutrient content and animal life.
628 Note prepared by CB Vand & Miljø. Sweden. Accessed on 10/01/2012.
629 URL: http://www.emarker.dk/pdf/Lake_Relief_test_2007_EN.pdf
- 630
- 631 Waterguru Bertram Case Study. No date. Accessed on 25/10/2011.
632 URL: www.waterguru.net.au/case_studies/bertram_lake_treatment.pdf.
- 633 Waterguru Case Studies. 2011. Accessed on 25/10/2011.
634 URL: www.waterguru.net.au/case-studies.
- 635 Waterguru LakeRelief. No date. Accessed on 25/10/2011.
636 URL: www.waterguru.net.au/products/LakeRelief.pdf.
- 637
- 638 Personal communications:
- 639
- 640 Lee, R. (01/10/2008) Personal communication. Australian Agent for Novozymes, Lee-
641 Stewart Pty. Ltd.
- 642 Lee, R. (01/02/2009) Personal communication. Australian Agent for Novozymes, Lee-
643 Stewart Pty. Ltd.

The tested bacterial-based bioremediation product did not actively lyse algal cells

The product suppressed algal growth in tanks over short periods (several days)

Testing product performance in tank-based experiments was hindered by limitations

No long-term nutrient immobilisation, thus unsuitable for natural wetland treatment

Table 1: Equipment list for physiochemical water analysis (used in all experiments in this study)

Parameter	Equipment
Turbidity	Hach Portable Turbidimeter 2100P
pH	Hanna Microcomputer pH meter HI 8424
DO	Hanna Microprocessor Dissolved Oxygen meter HI 9145
Temperature	Hanna Microprocessor Dissolved Oxygen meter HI 9145
Conductivity	Hanna Conductivity meter HI 8733
COD	Hach Portable Data-logging Spectrometer DR 2010
Chlorophyll <i>a</i>	Spectrophotometer Novaspec II
Fluorescence	Turner Design Fluorometer Model 10-005

Table 2: Limitations in tank experiments and their consequences

Nutrient and carbon limitation	Bacterial growth was impeded by unfavourable nutrient and carbon supply ratios available at times of product dosing.
Physiochemical extremes	As strong algal photosynthesis occurred throughout the whole reactor, pH and DO raised to extreme levels, hostile for bacteria.
Uniform conditions	In the confined reactor space, pH and DO concentrations were homogenous. No buffer zones existed, where conditions are less extreme.
Water mixing from aeration	Water motion influenced algal species composition and promoted certain - mobile - species (cyanobacteria) over others (green algae).
Surface to tank volume ratio	Measurements of chlorophyll <i>a</i> and turbidity were strongly influenced by the rate of algal attachment onto reactor surfaces.
Light intensity	Light regime in the reactors affected algal responses and possibly, bacterial/algal interaction.
Absence of food web	While some nutrients appeared to be assimilated into bacterial biomass, fast re-release seemed to occur in the tanks, intimately linked to bacterial biomass die-off conditions. This was related to the absence of a trophic cascade and thus, no immobilisation and transport of nutrients into higher organisms occurred.

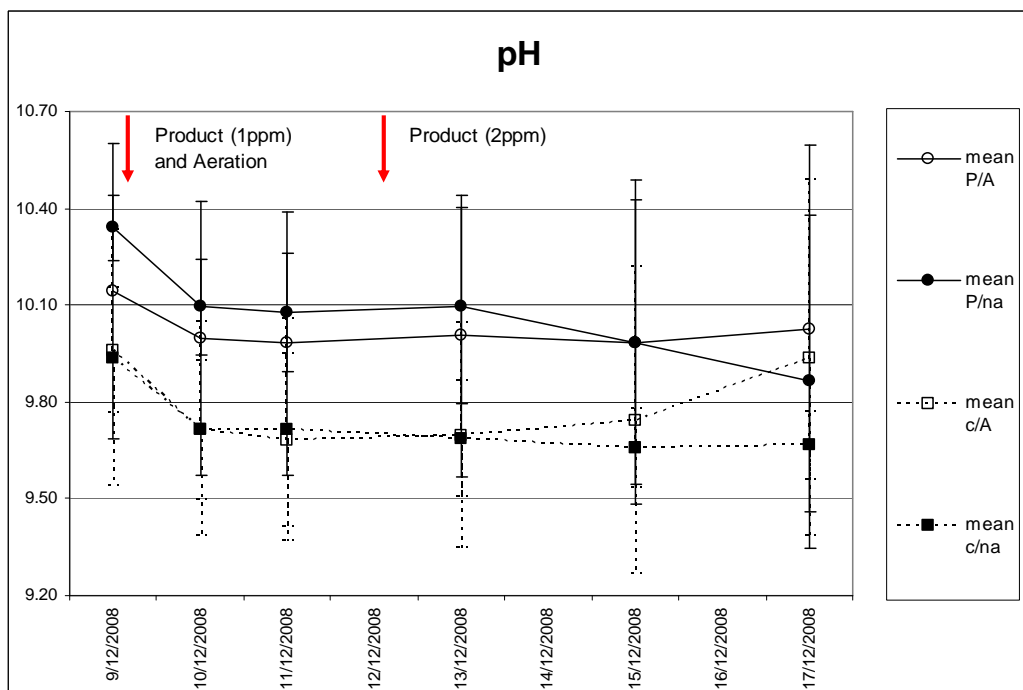


Figure 1: Testing algal lysing properties - pH value of the water column in product and control tanks (P/A = microcosms with product and aeration, P/na = microcosms with product and no aeration, c/A = microcosms with no product (control) and aeration, c/na = microcosms with no product (control) and no aeration). Red arrows indicate product application.

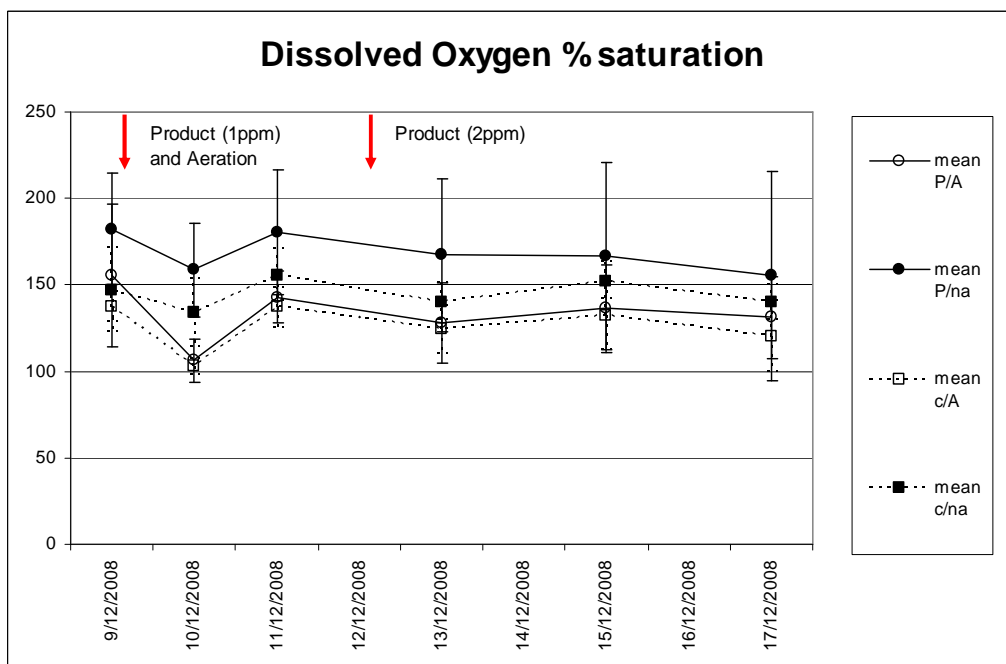


Figure 2: Testing algal lysing properties - DO saturation of the water column in product and control tanks

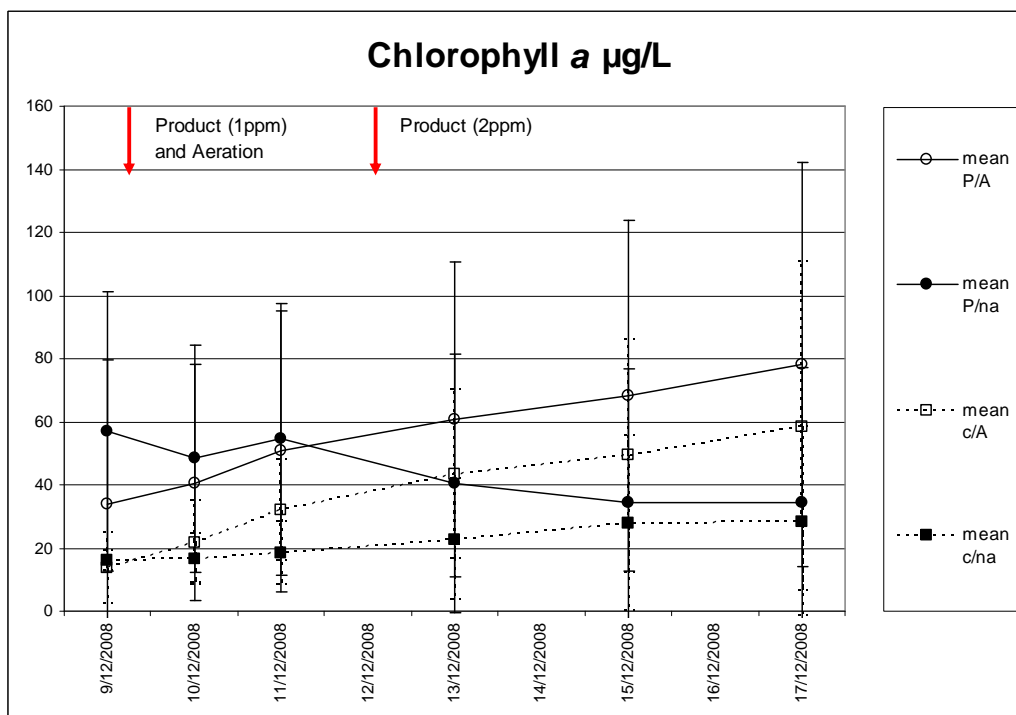


Figure 3: Testing algal lysing properties - Chlorophyll *a* concentration of the water column in product and control tanks

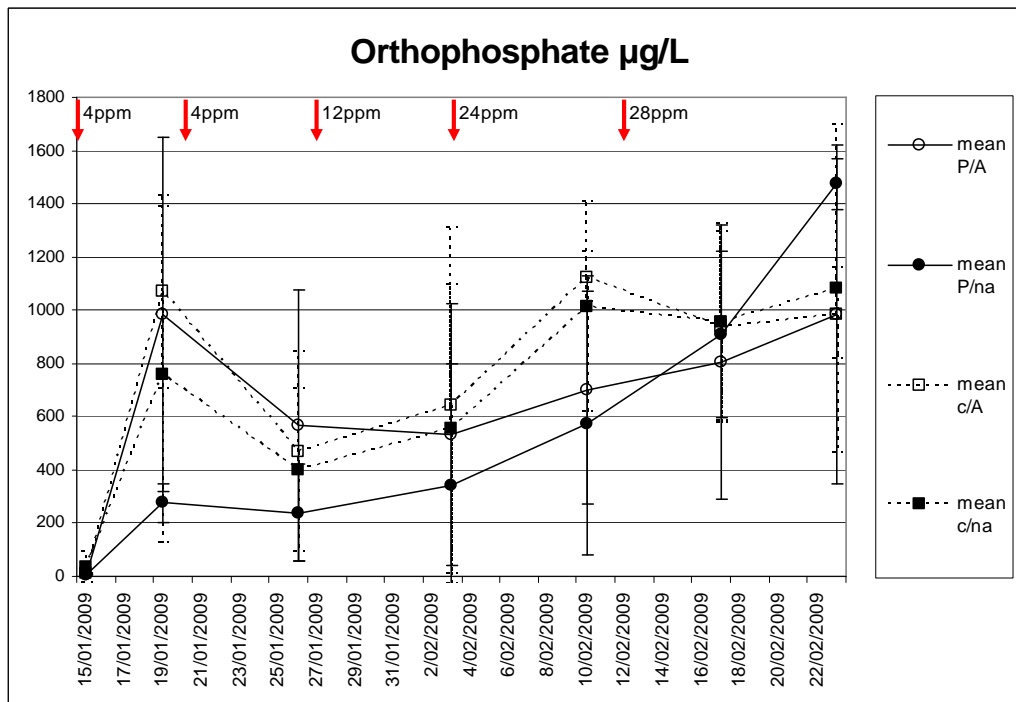


Figure 4: Testing nutrient competition - Orthophosphate concentration of the water column in product and control tanks.

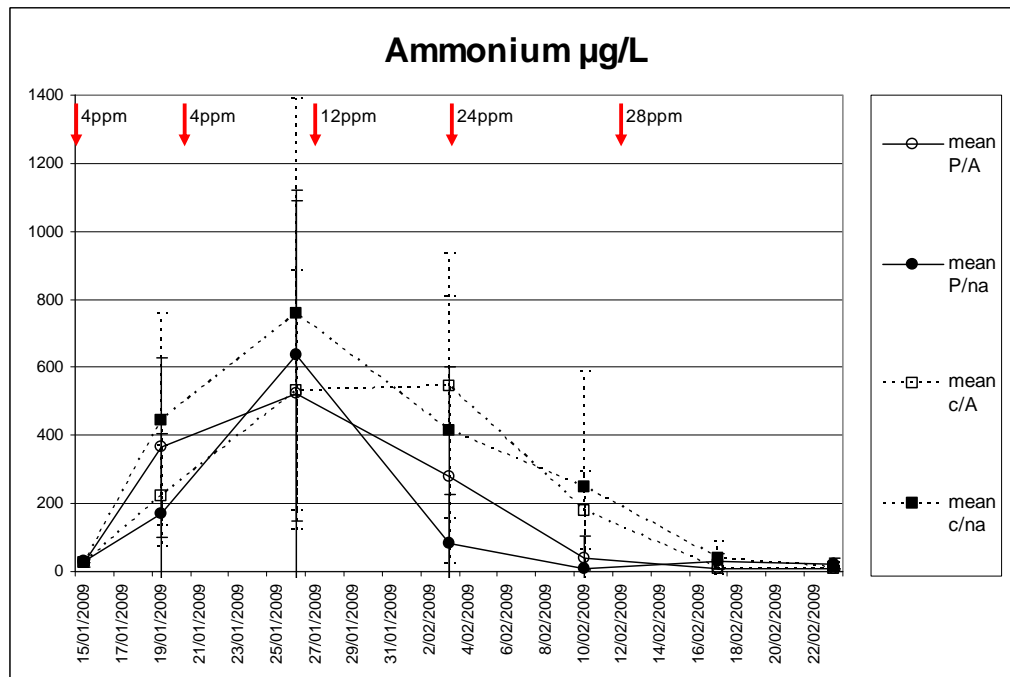


Figure 5: Testing nutrient competition - Ammonium concentration of the water column in product and control tanks.

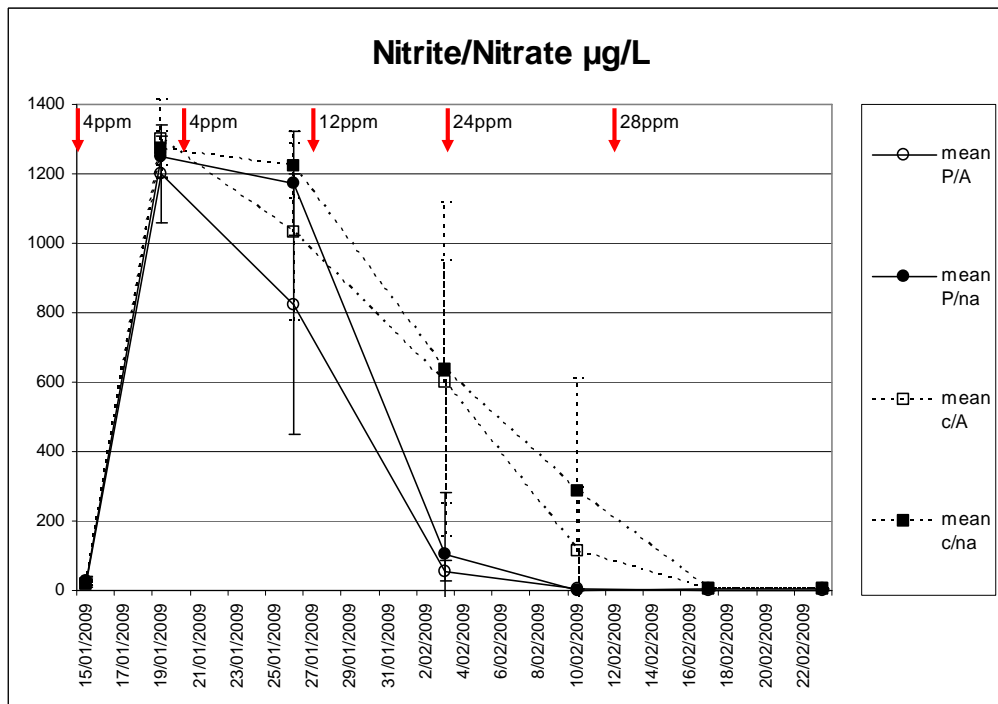


Figure 6: Testing nutrient competition - Nitrite and Nitrate concentration of the water column in product and control tanks

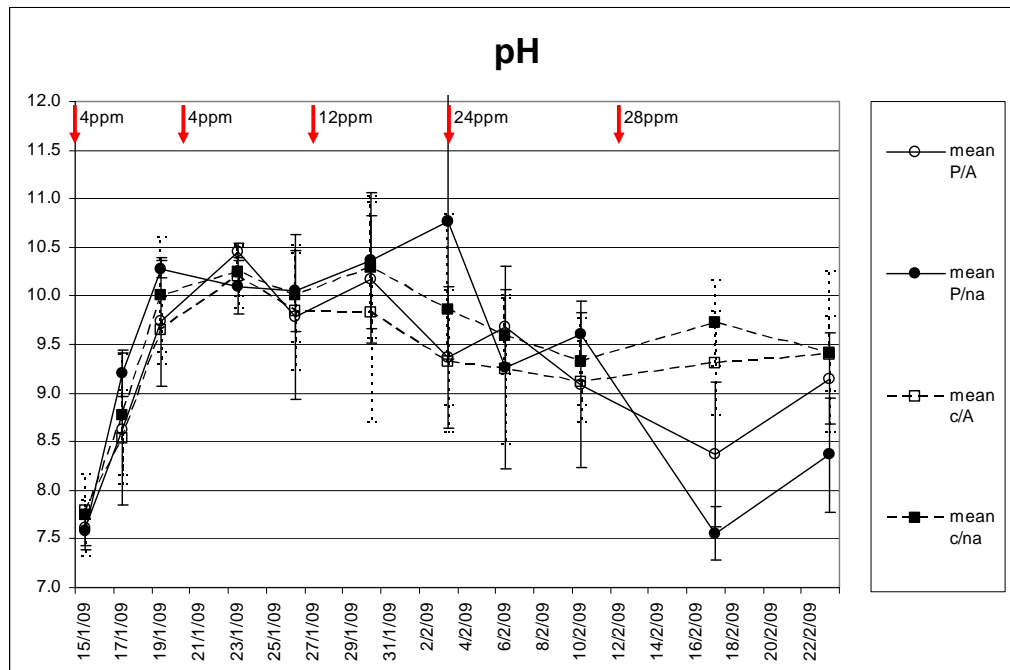


Figure 7: Testing algal growth suppression - pH value of the water column in product and control tanks

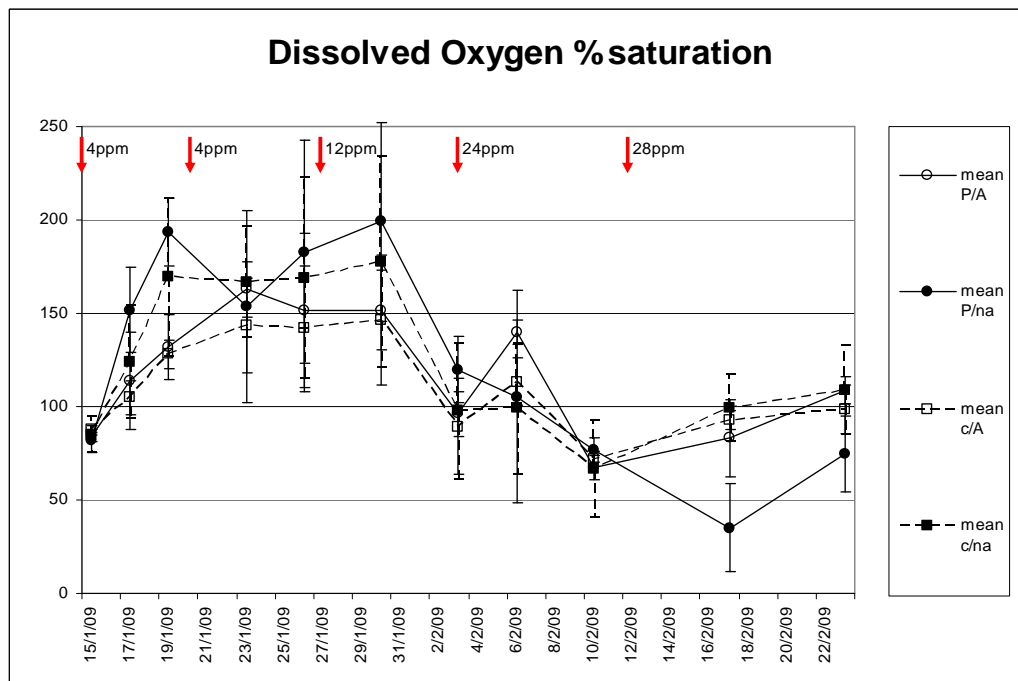


Figure 8: Testing algal growth suppression – DO saturation of the water column in product and control tanks

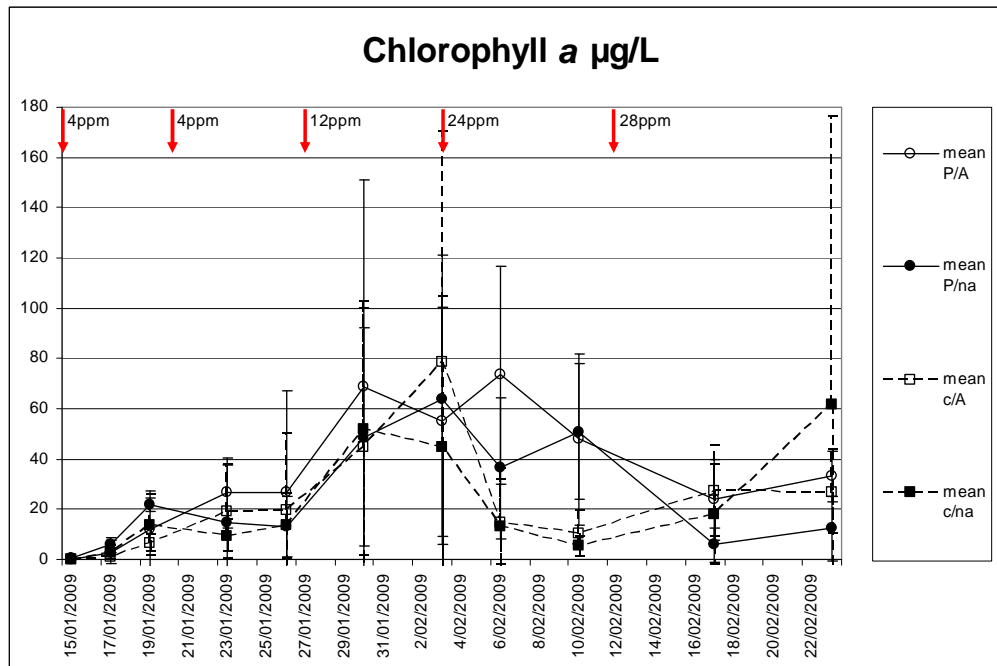


Figure 9: Testing algal growth suppression – Chlorophyll *a* concentration of the water column in product and control tanks